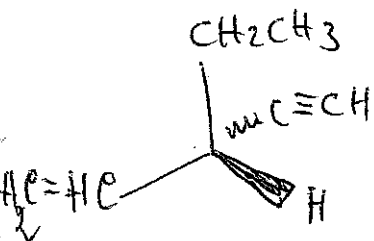
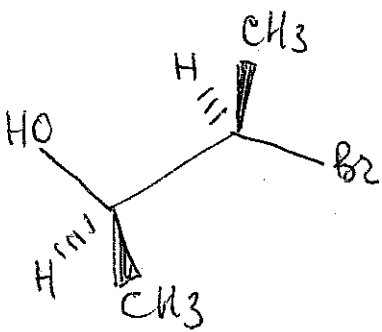
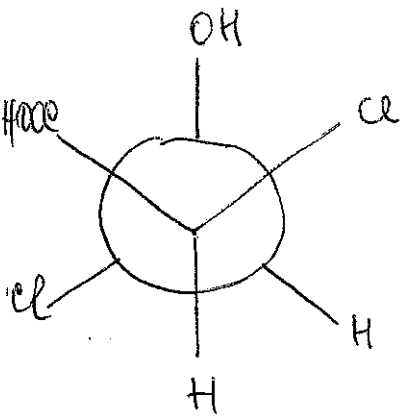
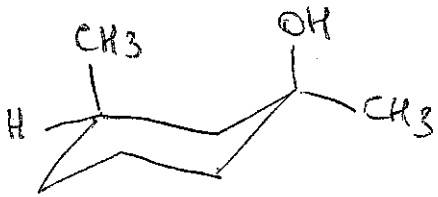
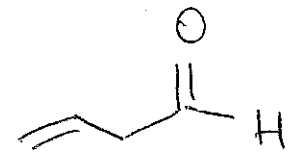
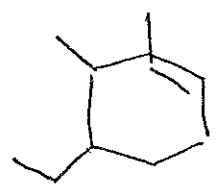
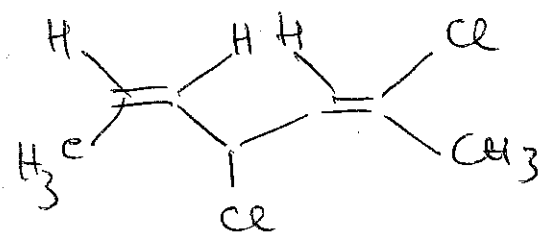
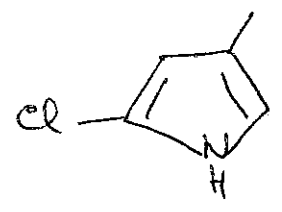
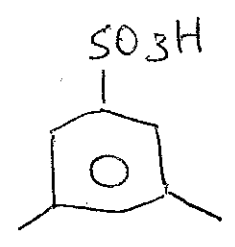
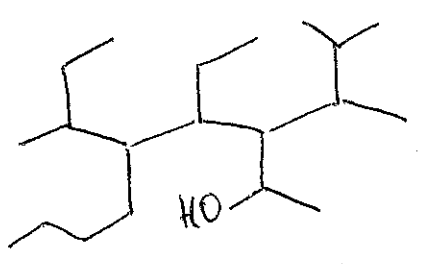


7.X.2013

1) Disegnare le proiezioni di Fischer e attribuire la configurazione assoluta



2) Scrivere il nome IUPAC e le notes. stereochimici, se opportuno, dei seguenti composti:



3) Individuare i composti aromatici motivandone la scelta:

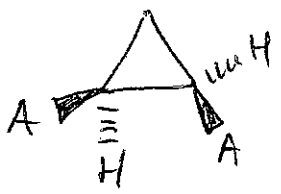
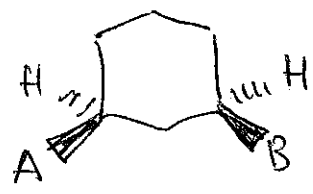
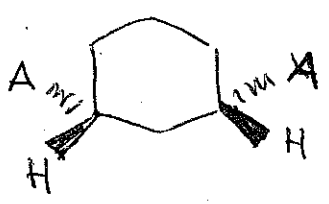
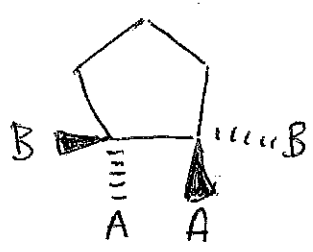
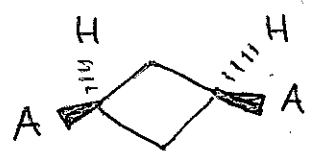


4) Per ciascun composto:

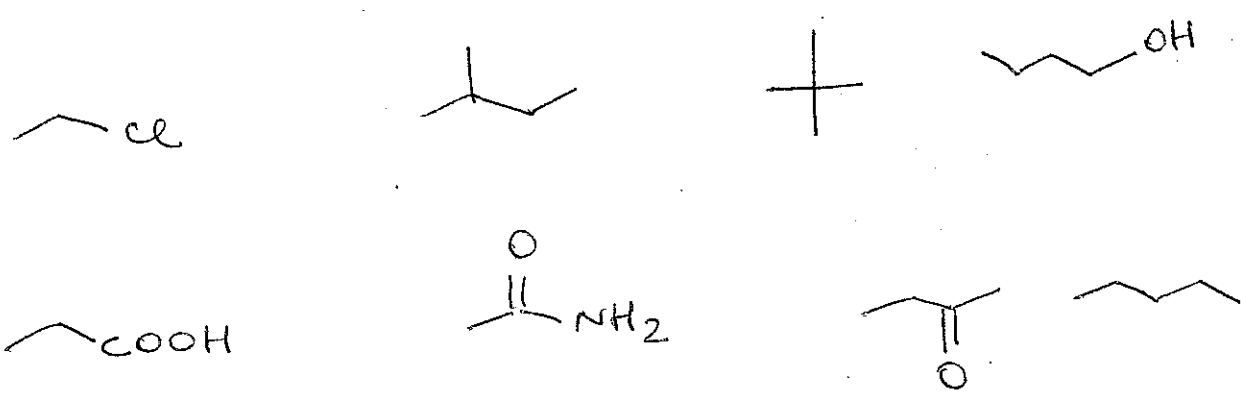
- dire se è chirale / achirale
- dire se può essere come coppia di enantiomeri.

In caso affermativo disegnare l'enantiomero

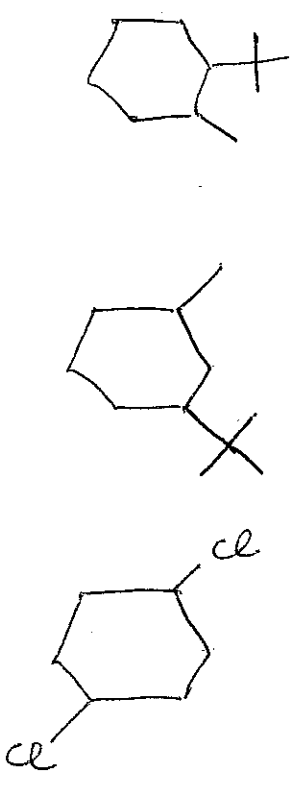
$A \neq B$



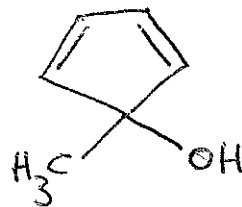
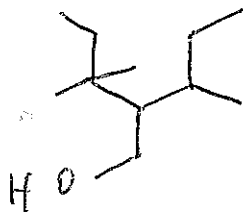
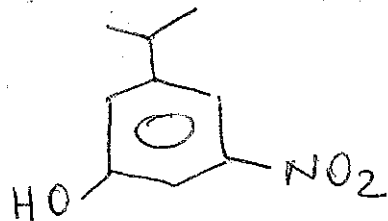
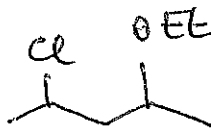
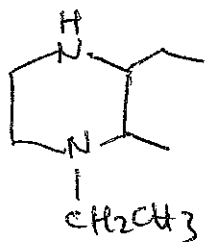
5) Disporre i composti in ordine di punto di ebollizione crescente motivandone la scelta (1=pto eboll. minore).



5) Disporre le conformazioni a sedia per l'isomero cis di ciascun composto indicando quella più stabile

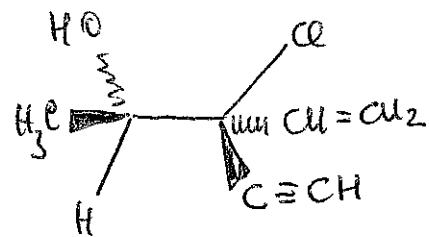
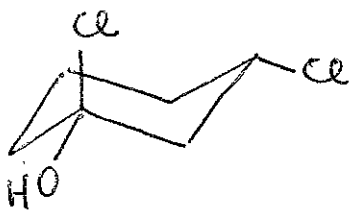
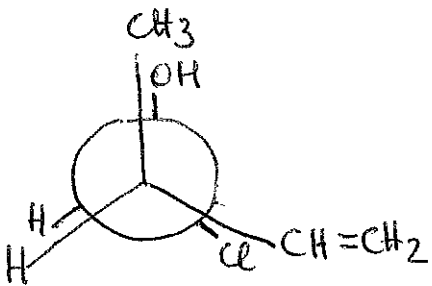
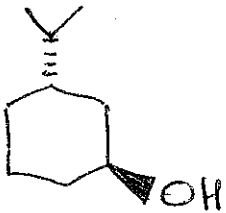


1) Attribuire il nome IUPAC e la notaz. stereoch. se opportuna:

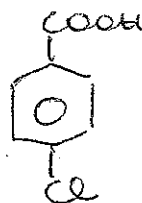
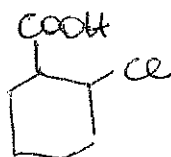
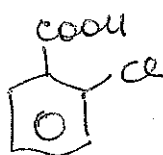
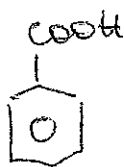
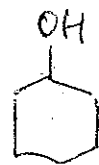
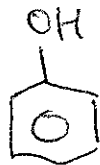
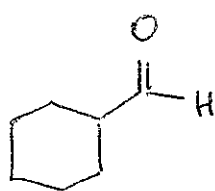
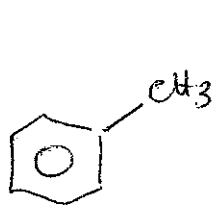


2) Descrivere l'ibridazione sp del carbonio (orbitali, geometria, ...) e indicare almeno due classi di composti che contengono C sp.

3) Disegnare le proiezioni di Fischer e attribuire la configur. assoluta:



4) Ordinare i seq. composti secondo l'acidità esercitata (1=ac. più debole) motivando la scelta e indicando l'idrogeno più acido di ciascun composto



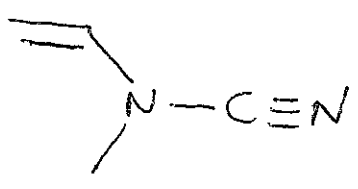
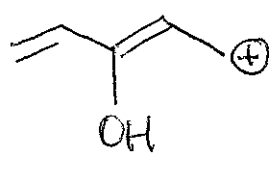
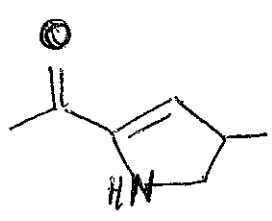
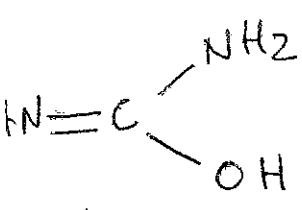
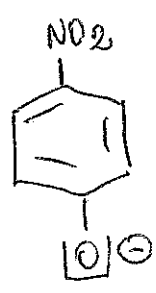
5) L'analisi polarimetrica di un composto puro mostra

$$[\alpha]_D^{25} = +24,2^\circ$$

Il composto:

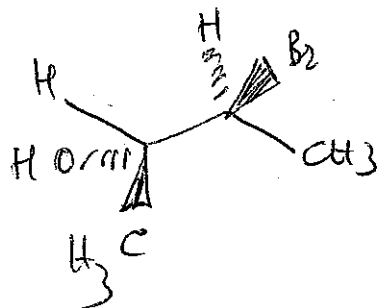
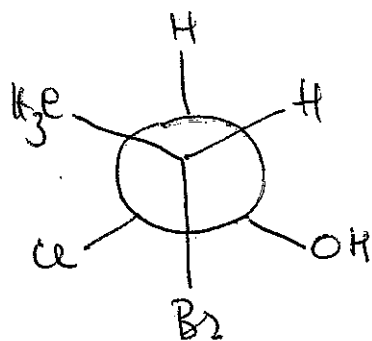
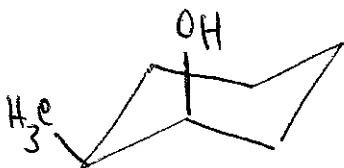
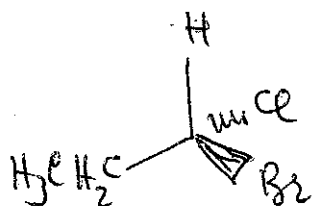
- può avere / ha / non ha 2 stereocentri
- può avere / ha / non ha un piano di simmetria
- può essere / è / non è chirale
- può avere / ha / non ha 1 solo stereocentro
- può essere / è / non è una forma meso

6) Scrivere le formule di risonanza!

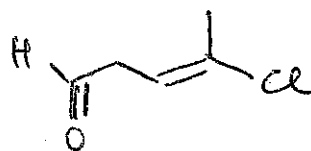
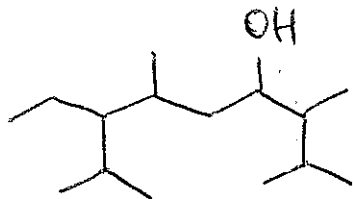
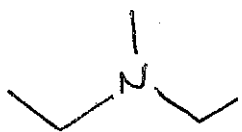
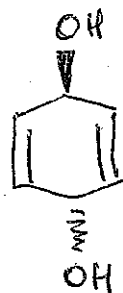
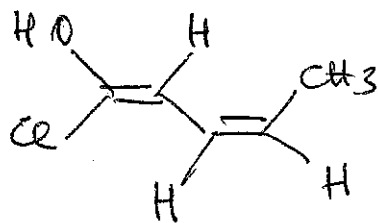
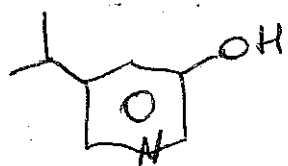


10/2/2014

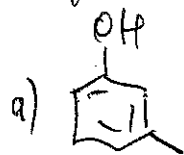
1. Disegnare le proiezioni di Fischer e determinare la sua configurazione assoluta



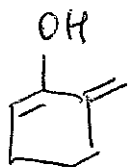
2) Atti bene il nome IUPAC e la notazione stereochimica, se opportuno



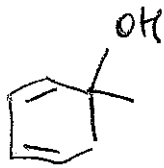
3) Per alcuni dei seg. composti è possibile l'esistenza di tautomeria cheto-enolica. Scrivere le 2 specie indicando quella prevalente



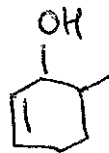
b)



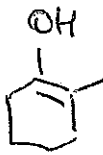
c)



d)



e)



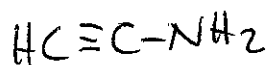
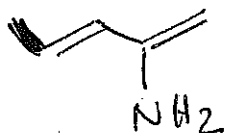
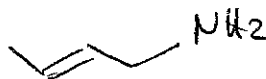
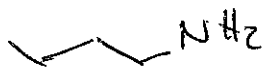
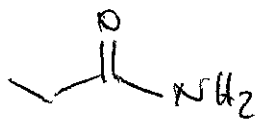
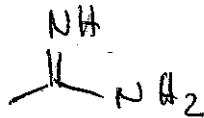
43) Utilizzando le formule appropriate rappresentare le conformazioni dei sep. composti attribuendo a ciascuna la stabilità relativa

cis-1,3-dibromocicloesano (x dice)

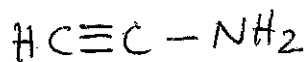
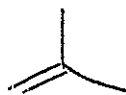
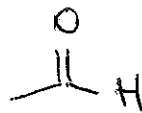
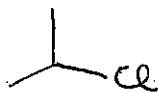
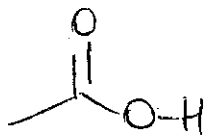
1,2-dicloroetano

(2E, 4Z) - 2,4-esadiene

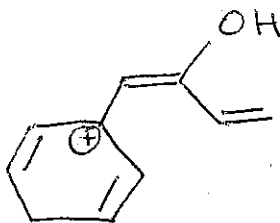
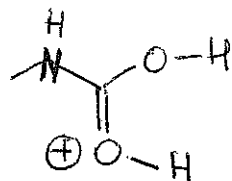
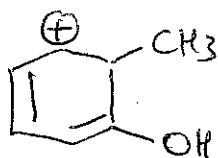
5) Ordinare i seg. composti secondo la basicità crescente motivandone la scelta: (1 = base + debole)



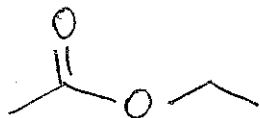
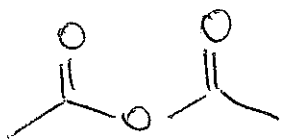
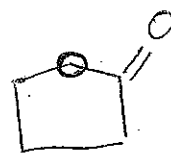
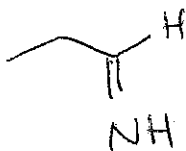
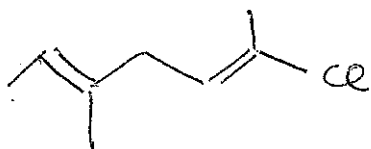
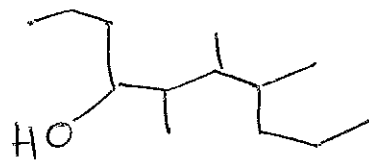
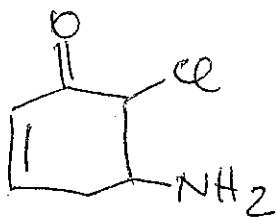
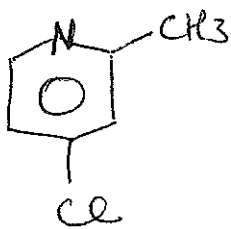
6) Abbassare il n. ossidazione ai vari atomi



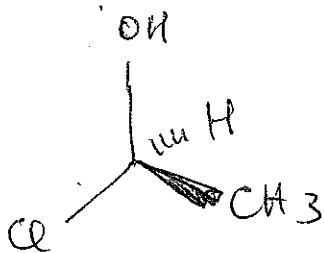
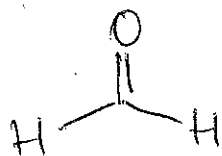
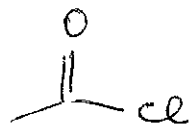
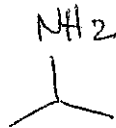
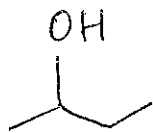
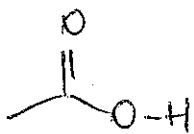
1) Scrivere le formule di risonanza dei seg. composti!



2) Scrivere il nome IUPAC e la nota stereoch., se opportuno, dei seguenti composti



3) Determinare il n. ossidazione dei vari atomi nei seq. composti:

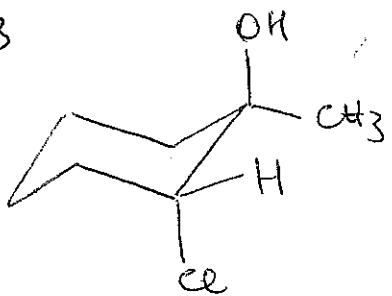
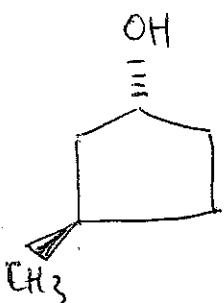
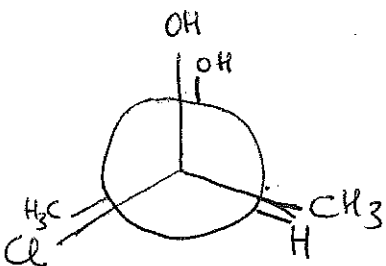
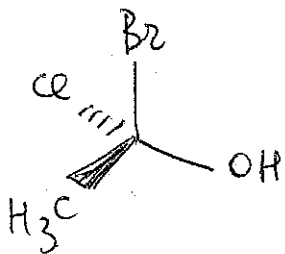


4) Scrivere le conformas. a sedia per i seg.
composti e per ciascuno composto attribuire
la stabilità relativa

(1S, 3R) - 1-bromo - 3-cloro - 1,3-dimetil cicloesano

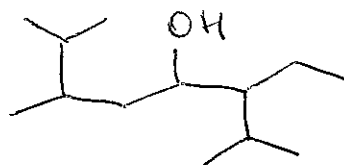
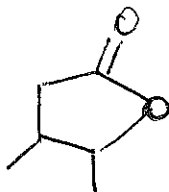
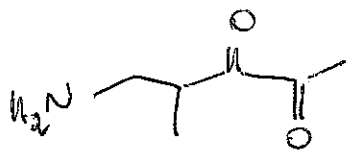
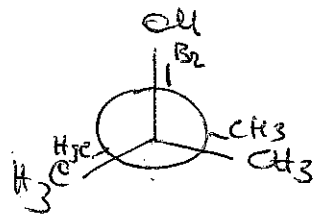
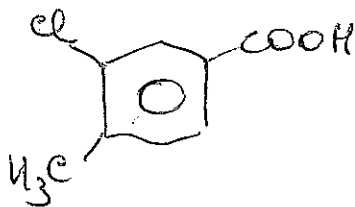
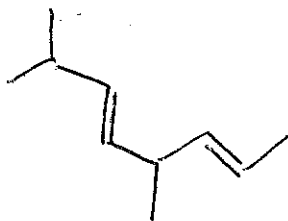
(1S, 2R) - 2-cloro - 1,2-dimetil cicloesano

5) Disegnare le proiezioni di Fischer e attribuire la configurazione assoluta:



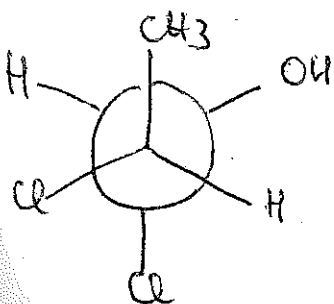
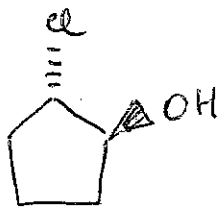
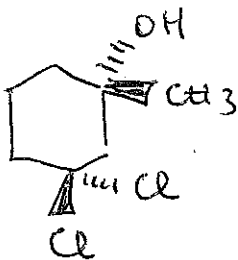
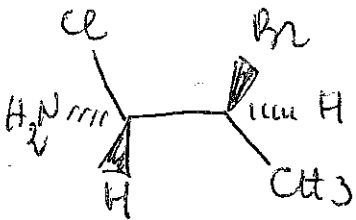
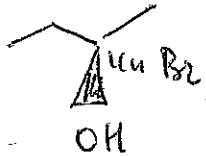
24/2/2014

1) Attribuire il nome IUPAC e le notes stereochimiche, se opportuno:

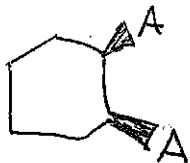
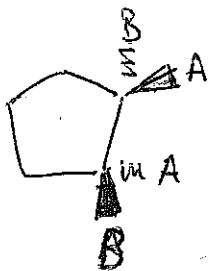
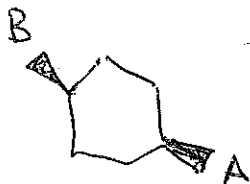
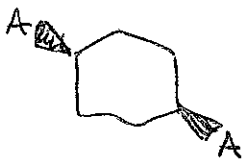
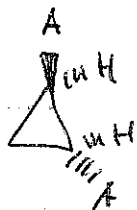
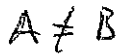


2) Descrivere l'ibridazione sp^2 del carbonio (orbitali, geometrie) indicando almeno tre classi di composti con C sp^2 .

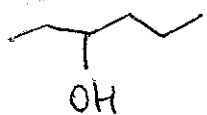
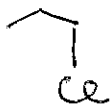
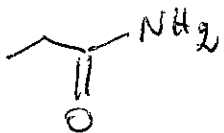
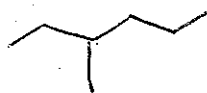
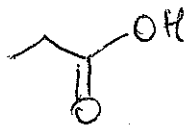
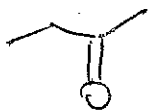
3) Si separare le proiezioni di Fischer e attribuire la config. assoluta?



- 4) Per ciascun composto a) dire se è chirale / achirale
b) se può esistere come coppia di enantiomeri,
~~non~~ dire pure l'enantiomero



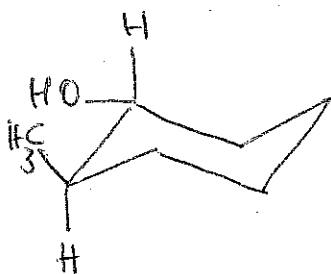
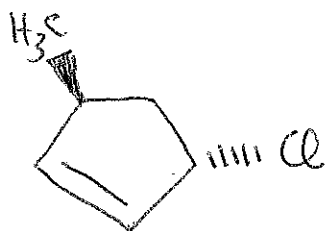
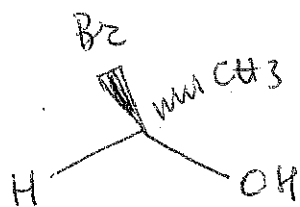
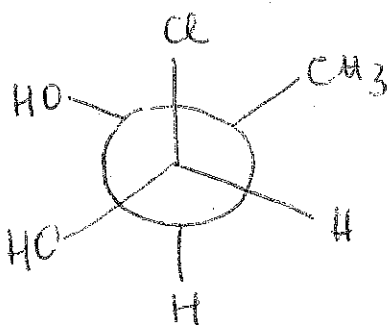
5) Disporre i seq. seguenti secondo pto ebolliz. crescente, motivandone le scelte. (1 = pto eb. minore)



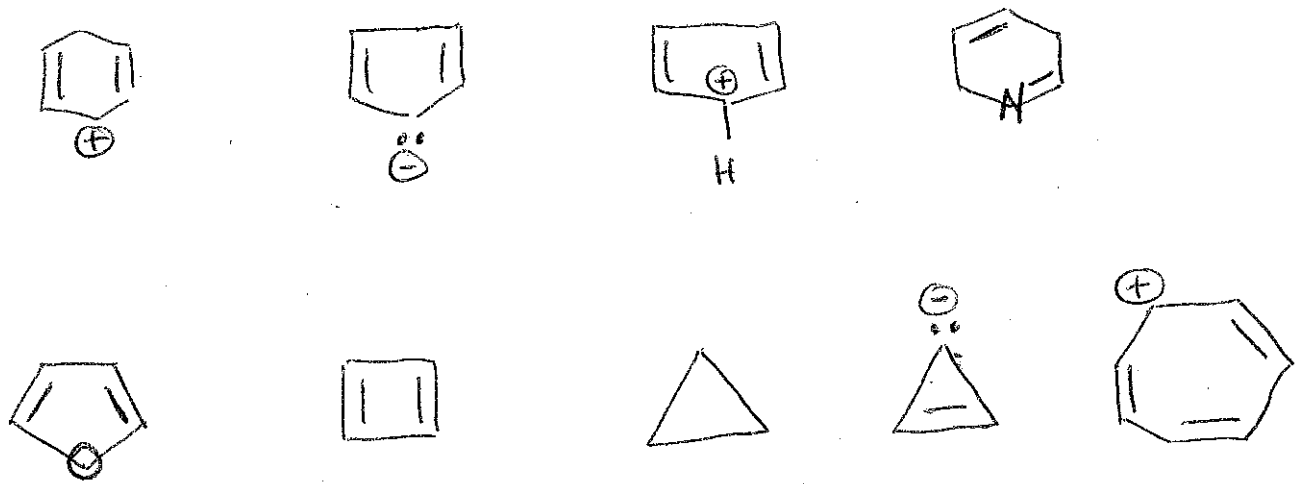
6) Individuare i conf. aromatici motivandone la scelta



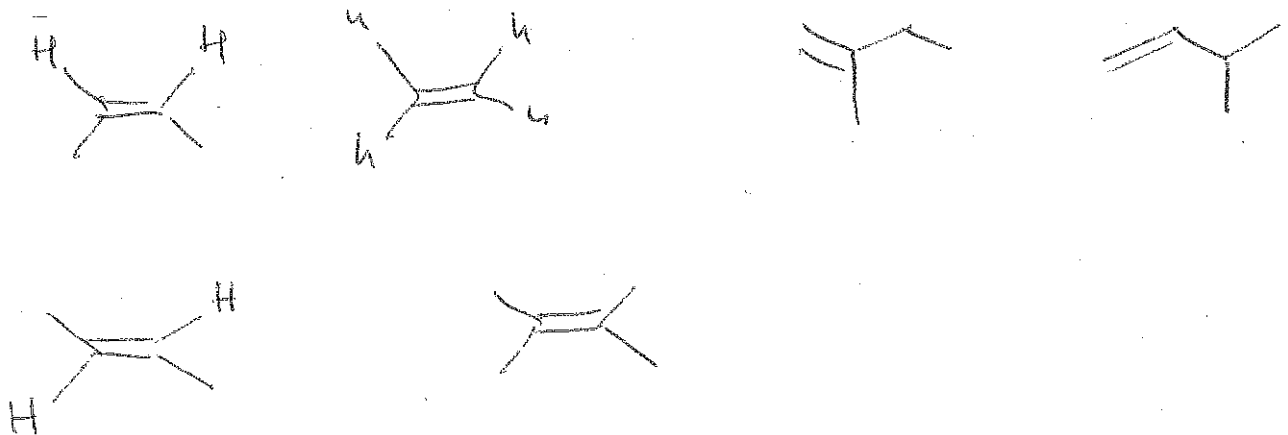
1) Disegnare le proiezioni di Fischer e attribuire le config. assolute ai sep. composti!



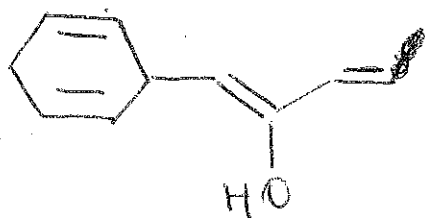
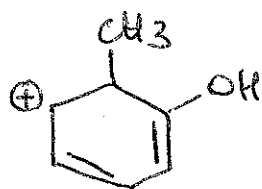
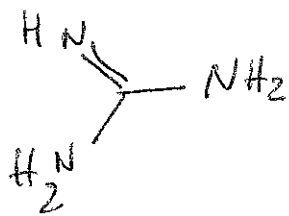
2) Individuare i composti aromatici motivandone la scelta:



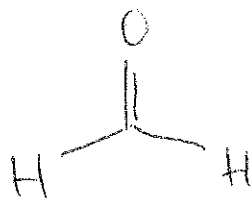
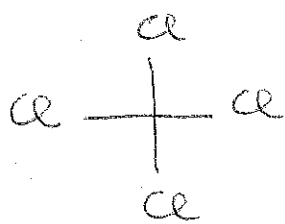
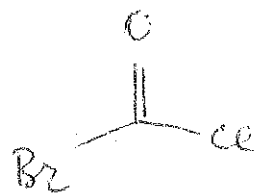
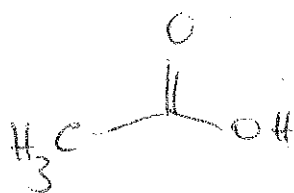
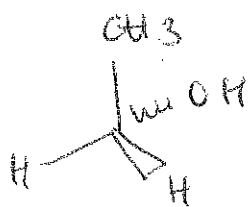
3) Disporre i sep. composti secondo la loro stabilità relativa (1 = più stabile) motivandone la scelta



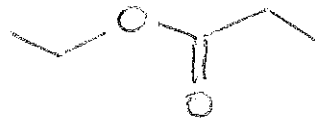
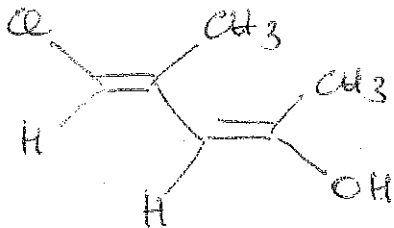
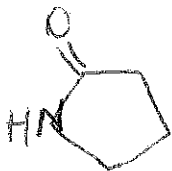
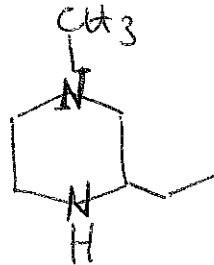
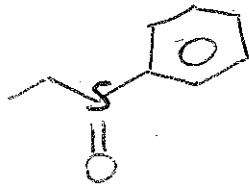
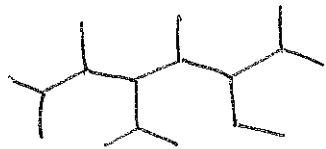
4) Rappresentare le strutture di risonanza per i seguenti composti



5) Determinare il n. ossidazione dei vari atomi:



6) Scrivere il nome IUPAC e la nota stereochimica, se opportuna, dei sep. composti:



UNIVERSITA' DEGLI STUDI DI SIENA

Corso di laurea in Scienze Chimiche

Compito di Chimica Organica - 1 modulo

17/7/2013

1. Stabilizzando le formule opportune scrivere:

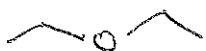
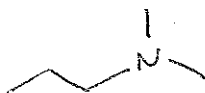
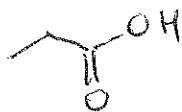
= un' ammina terziarie chirale.

= un sale di ammonio chirale

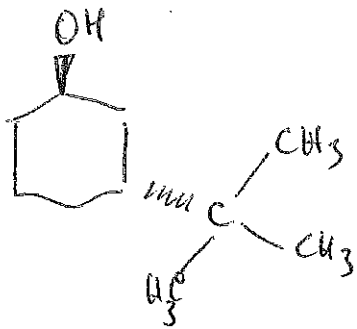
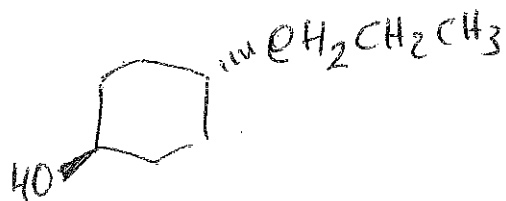
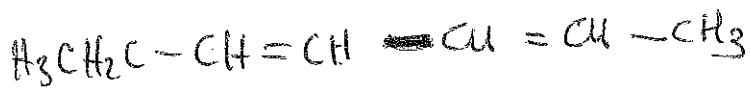
= un enantiomero del 2,3-diamminobutano

= due D-chetopentosi

2) Ordinare i sep. esempiti secondo pto ebollizione crescente (1= pto eb. minore) motivandone la scelta

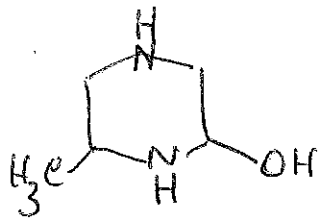
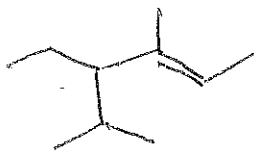
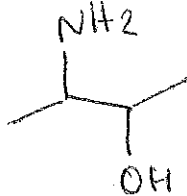
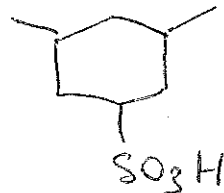
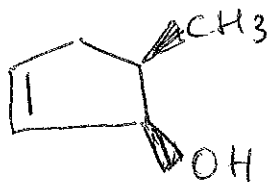


3) Per ciascun composto di separare le conformazioni più stabili (almeno 2) e attribuire la stabilità relativa. Per i composti aciclici considerare il legame evidenziato

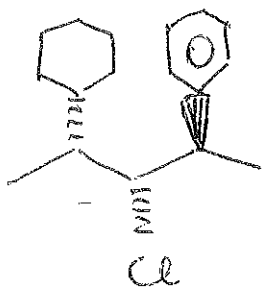
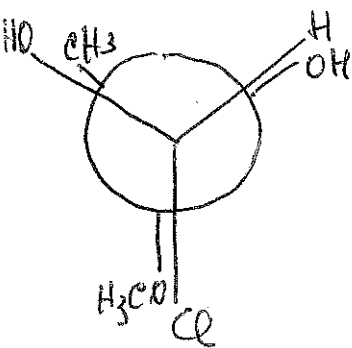
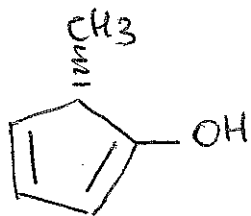


4) Scrivere il nome IUPAC e la nota. stereoch.

se opportuno:



1) Fare le proiez. di Fischer e attribuire la config. assoluta



2) Discutere l'equilibrio conformazionale del
(1R, 3S) - 3-metilcicloesano